

1 Flow Lattice Model for the Simulation of Chemistry Dependent 2 Transport Phenomena in Cementitious Materials

3 Hao Yin* Antonio Cibelli† Susan-Alexis Brown* Lifu Yang‡ Lei Shen§
4 Mohammed Alnaggar¶ Gianluca Cusatis* Giovanni Di Luzio‡

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6 Abstract

7 This study presents the formulation and validation of a three-dimensional (3D) Flow Lattice Model
8 (FLM) with application to the Hygro-Thermo-Chemical (HTC) model for analysis of moisture transport
9 and heat transfer in cementitious materials. The FLM is a discrete transport model formulated in
10 association with meso-mechanical models, such as the Lattice Discrete Particle Model (LDPM). This
11 enables the simulation of transport phenomena at the length scale at which the material exhibits intrinsic
12 heterogeneity. The HTC theoretical formulation is based on mass and energy conservation laws, written
13 using humidity and temperature as primary variables, and taking into account explicitly various chemical
14 reactions, e.g. cement hydration and silica fume reaction, as internal variables. In this work, the HTC
15 formulation was extended to include the effect of temperature on the sorption isotherm. The FLM
16 solutions were compared with those of a continuum finite element implementation of the HTC model and
17 experimental data available from the literature; the overall agreement demonstrates the reliability of the
18 proposed approach in reproducing phenomena such as cement hydration, self-desiccation, temperature-
19 dependent moisture drying, etc.

*Northwestern University, Department of Civil and Environmental Engineering, Evanston, Illinois, 60208, United States

†Politecnico di Milano, Department of Civil and Environmental Engineering, Milan, 20121, Italy

‡Hunan University, College of Civil Engineering, Changsha, 410082, China

§Hohai University, College of Water Conservancy and Hydropower Engineering, Nanjing, 210098, China

¶Oak Ridge National Laboratory, Oak Ridge, Tennessee, 37830, United States

‡Corresponding author: giovanni.diluzio@polimi.it

20 **Key words**

21 Discrete modeling, Moisture transport, Heat transfer, Cement hydration, Dual lattice

22 **1 Introduction**

23 The use of concrete as a building material is fundamental to the contemporary construction indus-
24 try. Its historical prevalence combined with present-day ease of production, low labor cost, and versatile
25 engineered properties has promoted concrete as an invaluable construction material. However, the con-
26 struction industry, and in particular concrete production, is responsible for up to 10% of total CO₂
27 emissions per year. Cement clinker alone generates 0.9 tons of CO₂ per ton of material [1]. Reducing the
28 resulting environmental effects is critical to safeguarding the future of construction. One factor which
29 will lessen the impact of any material is improving the durability of the structure. Extending the service
30 life time of concrete structures will minimize the need for new material production, and thus not only
31 bring economic benefits but also reduce carbon emissions.

32 The durability of concrete structures is a function of a complex coupling between multi-physical
33 phenomena, including the chemical aging process of the material, ingress of various deteriorating agents,
34 and mechanical damage due to loads [2]. While design codes typically consider the suitability of the
35 structures to sustain predicted loads at the time of construction, the long-term behavior of the structure
36 may not be appropriately predicted, resulting in additional material and economic costs, and often
37 compounding the initial carbon emissions. It is, therefore, necessary to adopt design principles which
38 take into account the many facets which influence material durability to produce a structure with the
39 optimal lifespan and thus minimize environmental impacts.

40 One of the primary causes of the deterioration of reinforced concrete is corrosion of the reinforcing
41 steel, which can reduce tensile capacity and lead to spalling of the concrete cover [3]. This consequently
42 shortens the service life of structures and increases maintenance costs. These effects are mitigated through
43 the use of high-performance concrete (HPC) mixtures. Developed in the early 90's, HPC is characterized
44 by low permeability and thus reduced risk of corrosion. Unfortunately, structures made from HPC are
45 typically more vulnerable to cracking, especially at very early ages. The benefits of low permeability are
46 completely lost when cracking initiates, as the cracks provide direct paths for corrosive agents to reach
47 the reinforcing steel.

48 The cause of this early-age cracking in HPC is primarily attributed to resistance to volume changes

49 resulting from: (1) thermal expansion due to heat released by exothermic chemical reactions (cement
50 hydration and pozzolanic reactions); (2) shrinkage due to water loss through exposed concrete surfaces
51 (drying shrinkage); and (3) shrinkage due to water consumption during cement hydration and other
52 chemical reactions (autogenous shrinkage). For normal strength concrete exposed to the environment,
53 drying shrinkage dominates, but in the case of exposed HPC, in which autogenous and drying shrinkages
54 combine, resulting in larger overall shrinkage [4].

55 As mentioned, these phenomena occur at early age, when the concrete strength and stiffness are still
56 under rapid development and are directly influenced by curing conditions, such as temperature and rela-
57 tive humidity. The resulting overall shrinkage cannot, therefore, be easily predicted solely from the mix
58 design. Thus, a reliable and predictive analysis of the concrete structure durability requires the formula-
59 tion of a consistent theoretical and computational framework that includes all of the various phenomena
60 characterizing the aging and evolution of concrete transport properties from early age. Therefore the
61 characterization of moisture content, temperature evolution and reaction of cementitious materials are
62 fundamental in this context.

63 In the literature, there are many experimental, theoretical, and computational studies on the behavior
64 of concrete at early age. Studies on the analysis of self-desiccation include [5, 6, 7, 8], or both self-
65 desiccation and drying such as [9, 10]. Many of the numerical models in the literature describe the
66 cement hydration processes in conjunction with hygro-thermal and mechanical considerations. Among
67 them, it is worth noting the dispersion model of [11] and the work of [12] with simple formulations for
68 hydration based on spatial considerations, as well as the models of [13] and [14] that describe cement
69 hydration based on the reaction kinetics and allow the evaluation of strength in time. In addition, [15]
70 proposed a simple model that can take into account the effect of temperature on strength evolution,
71 [16] developed a mesoscopic model for temperature and hydration degree in early-age concrete, and [17]
72 proposed a simplified model to predict the internal relative humidity in early-age concrete under different
73 curing humidity conditions.

74 More complex models allow the description of hydration, heat diffusion, moisture transport, chemical
75 shrinkage, and self-desiccation [18, 19, 20, 21, 22, 23]. Among them, the Hygro-Thermo-Chemical (HTC)
76 model proposed by [24, 25] formulates the evolution of cement hydration and of pore relative humid-
77 ity in concrete, based on moisture transport and heat transfer governing equations. The model uses
78 phenomenological evolution laws to describe the reaction degrees for cement and silica fume, and the as-
79 sociated changes in evaporable and chemically bound waters, along with adsorption/desorption isotherms

80 and permeability of concrete. The capability of the HTC model has been demonstrated through exten-
81 sive numerical studies by many authors [25, 26, 27, 28, 29, 30, 31, 32]. Recently, a multiscale approach,
82 entitled the ONIX model, has been proposed in [33], where all the parameters governing the HTC con-
83 stitutive relations are identified by using the output of micro-scale numerical simulations carried out by
84 the CEMHYD3D model [34, 35].

85 The influence of cracking on the mass transport, which governs many engineering applications of
86 construction materials, is still a challenge from the numerical modeling point of view and different ap-
87 proaches are reported in the literature. Some models are based on continuum mechanics [36, 31], while
88 others combine continuum mechanics with discrete representation of cracks [37, 38, 39, 40]. Conversely,
89 discrete approaches, such as discrete element method, lattice and network models, have been formulated
90 to simulate cracking, mass transport and their coupling, as in [41, 42, 43, 44, 45, 46, 47, 48, 49]. It is
91 worth mentioning one network approach based on the Delaunay tessellation of a random set of points [48,
92 49]. In this approach, the physical transport processes have been first simulated by a multi-dimensional
93 network of 1D elements placed on the Delaunay edges, and the element properties are determined by
94 the corresponding Voronoi tessellation [48]. In a later study, the transport elements were placed on the
95 Voronoi edges (i.e. a dual network), maintaining the structural elements on the Delaunay edges [49]. With
96 this approach, the crack effects on diffusion processes are more physically reproduced, as the transport
97 elements aligned with the crack directions. However, in this method, as well as in most other models in
98 the literature, the discrete feature of the model is not related to the actual heterogeneity of the material.
99 This feature is critical when transport processes are considered in cementitious heterogeneous materials,
100 as demonstrated for chloride transport in concrete, particularly in non-saturated conditions, see [50].

101 The Lattice Discrete Particle Model (LDPM), originally proposed by Cusatis et al. [51, 52], is a
102 discrete model simulating concrete material heterogeneity at the mesoscale (coarse aggregate) and has
103 proven its capability in accurately simulating many mechanical behaviors, especially the mesoscale failure
104 behaviors of concrete materials [53, 54, 55, 56, 57, 58, 59, 60]. Some explorations of the capability of
105 the LDPM in solving multiphysics problems have also been performed recently [61, 62, 63, 64, 50, 65].
106 The Multiphysics Lattice Discrete Particle Model (M-LDPM) is formulated and fully integrated with the
107 existing LDPM to extend the capability of LDPM for solving multiphysics problems, such as moisture
108 diffusion, heat transfer, chloride diffusion, and crack healing within the LDPM framework [61, 62, 63, 64,
109 50, 66].

110 The objective of this study is to present the first step towards the formulation of the 3D Flow Lattice

111 Model (FLM) with application to the HTC model. The FLM approach forfeits the continuum assumption
112 by way of discrete modeling at the mesoscale from the LDPM geometry. This means that FLM can more
113 accurately consider damage and its localization in cracks by discontinuities between adjacent cells and it
114 is able to reproduce critical features of material heterogeneity being 3D flow network based on the LDPM
115 geometry. This work, which focuses only on the first point, presents the new formulation of the 3D
116 FLM which represents an important development in the field of concrete construction in which reliable
117 predictive numerical model of the chemo-transport phenomena are fundamental to design durable and,
118 therefore, sustainable structures and infrastructures. Also, the quality of the FLM results are verified by
119 comparing the FLM solutions with the finite element solutions since both of them must converge to the
120 analytical solution when the size of the discretizing mesh decreases and same constitutive laws with same
121 parameters are employed.

122 In the following paper, the balance equations of moisture transport and heat transfer are derived in the
123 discrete context, and the theoretical HTC formulation is outlined. Then, the numerical implementation
124 is presented and validated using a selection of experimental data available in the literature. For the first
125 time, the effect of the temperature on the moisture diffusion is also verified. The FLM full coupling with
126 LDPM, i.e. effects of damage, micro-cracks, and cracks, as well as the updates and improvements of the
127 HTC formulation at the mesoscale are behind the scope of the manuscript and are left for future work.

128 **2 The Flow Lattice model**

129 **2.1 The Lattice Discrete Particle Model (LDPM) tessellation and the Flow** 130 **Lattice Model (FLM) network**

131 A pair of topologically dual and coupled lattice systems are involved in LDPM. One is referred to
132 as the LDPM cell system, which provides a geometrical characterization for mechanical behaviors of
133 concrete. The other is referred to as the *Flow Lattice Model* (FLM) system - the key focus of this work -
134 which provides the geometrical characterization of the flow (diffusive) behaviors of concrete.

135 The geometric representations of the discrete models, including LDPM and FLM, are crucially im-
136 portant for representing the model behaviors with high fidelity. To capture the granular nature of the
137 concrete internal mesostructure, LDPM discretizes the model domain with randomly packed polyhedral
138 cells, which represent the coarse aggregate particles embedded in a cementitious fine mortar (hereinafter
139 referred as LDPM tessellation). The cell size corresponds to the length scale at which the significant

140 material heterogeneities (coarse aggregates fields) are observed. The construction of the dual lattice
 141 systems is briefly summarized hereinafter, and the details of the tessellation can be found in [51]. (i)
 142 Idealized spherical particles that approximate the aggregate particles of concrete are randomly packed in
 143 the model domain without overlapping. The particle size distribution curve follows a Fuller sieve curve to
 144 reproduce realistic concrete mesostructures. Over the model boundaries, nodes (equivalent to zero-radius
 145 aggregates) are randomly placed. A typical aggregate distribution is shown in Fig. 1a. (ii) Given the
 146 particle centers, a constrained Delaunay tetrahedralization [51] is performed to divide the domain volume
 147 with a mesh of tetrahedra (LDPM tets, as shown in Fig. 1b). In each LDPM tet, 12 LDPM facets are
 148 constructed by sequentially locating edge points E_{ij} , face points F_k , and a tet point N and forming
 149 small triangular facets " $NE_{ij}F_k$ ". (explain here face/edge/tet point) The construction of LDPM facets
 150 can be illustrated as follows (see also [51]): edge-points are defined at midways of the edges belonging
 151 to the associated particles (E_{12} for P_1 and P_2 in Fig. 1b). Regarding the face-points, on each face of
 152 the tetrahedron, the midway point (for example, F_4 for face 4) located on the straight lines connecting
 153 each face vertex to the edge-point located on the edge opposite to the particle under consideration are
 154 identified as face points. Regarding the edge-points, these points locate at midway of the line counterpart
 155 not belonging to the associated particles. In Fig. 1b, the point F_4 associated with vertex P_3 and edge-
 156 point E_{12} is shown. The tet-point is defined as the centroid identified on the straight lines connecting
 157 each vertex of the tetrahedron with the face-point on the face opposite to the vertex under consideration
 158 and located at midway of the line counterpart not belonging to the associated particle. The tet-point N
 159 associated with vertex P_4 is shown in the Fig. 1b. Finally, a LDPM tet with its all 12 LDPM facets is
 160 shown in Fig. 1c.

161 (iii) The model domain is then subdivided through these LDPM facets into a system of polyhedral
 162 cells. The surfaces of the polyhedral cells (consisting of LDPM facets, as shown in Fig. 1d) define the
 163 potential failure locations. The prerequisite of this statement is that, the LDPM mesoscale geometry
 164 in which polyhedral cells containing each particle are assumed rigid and those rigid discrete cells are in
 165 contact through the facets in which stress and strain are defined and where the deformation takes place.
 166 Therefore, the surface of the polyhedral cells (LDPM facets) represents the potential failure location, i.e.,
 167 where deformation and cracking can happen (the cell is rigid and can't deform). The LDPM cells and
 168 the tessellated model domain are shown in Fig. 1e and f, respectively.

169 (iv) With the tessellated domain, a flow network can be formed by connecting the tet points belonging
 170 to each pair of two adjacent LDPM tetrahedra (Fig. 1g) with 1D line elements (e.g., N_1N_2 line segment

171 in Fig. 1h), the resulting network is shown in Fig. 1i.

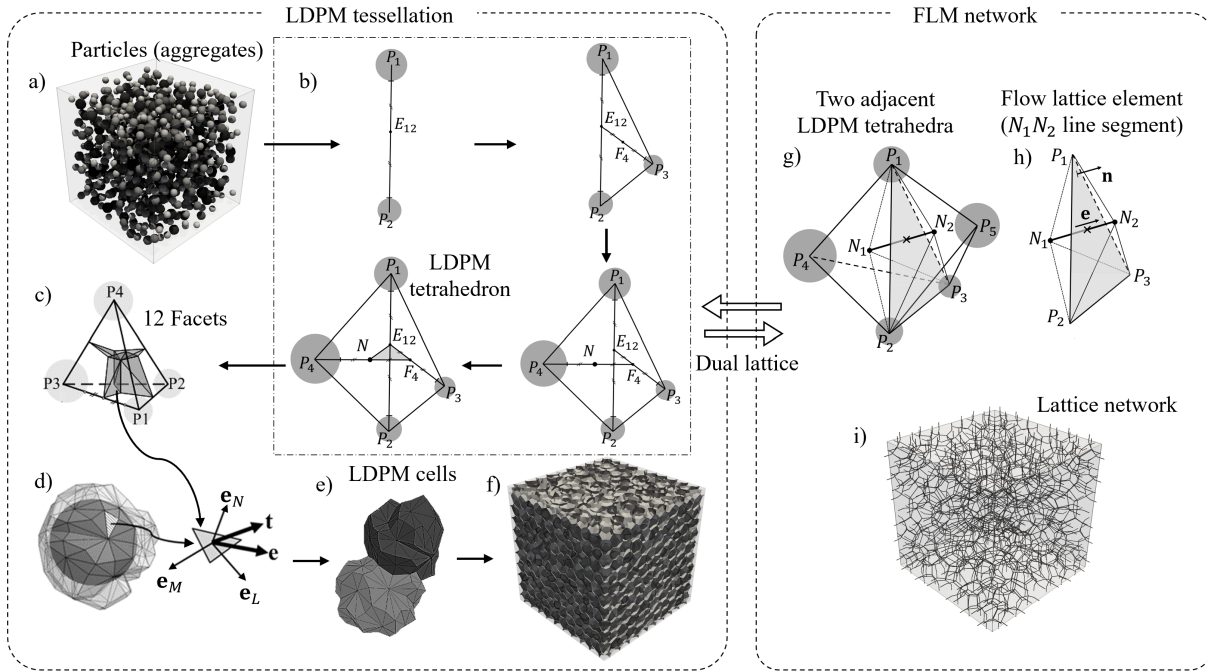


Figure 1: Lattice Discrete Particle Model (LDPM) tessellation and *Flow Lattice Model* network: a) idealized particles (aggregates) in the specimen, b) the construction of the LDPM tetrahedron (tet) from the creation of edge points E_{ij} , to that of face points F_k and tet points N , and finally to the formation of triangular facets " $NE_{ij}F_k$ ", c) LDPM tet element with 12 facets, d) a particle surrounded by associated LDPM facets, e) two adjacent LDPM cells, f) a typical LDPM tessellated specimen, g) two adjacent LDPM tetrahedra and their tet points N_1, N_2 , h) the associated *Flow Lattice Element* N_1N_2 , and i) the flow lattice network

172 The basic unit of the FLM network is called a *Flow Lattice Element* (FLE), where each FLE can be
 173 seen as a small conduit across the boundary of two adjacent LDPM tets (Fig. 2a). The properties of each
 174 FLE are characterized by the geometries of the associated tetrahedra, and potentially by the relative
 175 motions (e.g. crack openings) of the pairs of tetrahedra. The domain volume (V) of a FLE consists of
 176 two pyramids, referred as "side N_1 " (V_1) and "side N_2 " (V_2). The segment length of FLE, l , can be
 177 decomposed into two segments at the intersection of line segment with the tetrahedron surface $P_1P_2P_3$
 178 (Fig. 2b). The segment lengths l_1 and l_2 associated with side N_1 and side N_2 can be defined by the length
 179 proportionality coefficients $0 \leq g_1 \leq 1$ and $0 \leq g_2 \leq 1$, which satisfy the relations $l_i = g_i l$ ($i = 1, 2$). As
 180 shown in Fig. 2b, \mathbf{n} is the unit normal vector of the surface $P_1P_2P_3$, and \mathbf{e} is the direction vector of FLE
 181 from N_1 to N_2 . The projected area of triangular cross-section $P_1P_2P_3$ (A_0) in the \mathbf{e} direction is defined
 182 as $A = |\mathbf{n} \cdot \mathbf{e}| A_0$. One can write $V = V_1 + V_2 = Al/3$, generally, the normal vector \mathbf{n} is not necessarily
 183 parallel to direction vector \mathbf{e} .

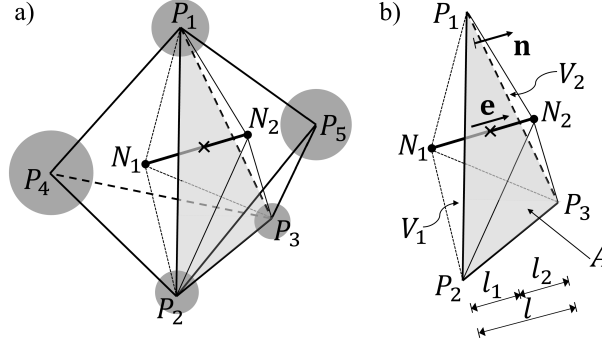


Figure 2: Diagrams of the *Flow Lattice Element* (FLE): a) FLE generation from the dual LDPM tetrahedra b) FLE geometry

184 Three types of FLE exist: regular (or type 1) elements - all interior elements belong to this category,
 185 as depicted in Fig. 3a; transitional layer (or type 2) elements, which corresponds to the elements having
 186 two nodes - one interior tet point and the other on the boundary of the whole model, as depicted in
 187 Fig. 3b; boundary layer (or type 3) elements, which shares the node on the boundary of a type 2 element,
 188 and then extends orthogonally to the outside of the model boundary. Different from type 1 and type
 189 2, the boundary layer element consists of two triangular prisms with equal volumes $V_1 = V_2$ and equal
 190 segment lengths $l_1 = l_2$, as depicted in Fig. 3c. Type 3 elements are used to enforce the boundary
 191 conditions; details will be discussed in following sections. The formulation of FLEs is the same for all
 192 types of elements, but the term values are governed by the geometric properties of different element types.

193 2.2 Balance equations for moisture transport and heat transfer in the Flow 194 Lattice Element (FLE)

195 For the transport problem of water mass and heat in concrete, the relative humidity h and temperature
 196 T are selected as the primary fields (a.k.a. the degrees of freedom in numerical analyses). The values at
 197 LDPM tet points N_1 and N_2 , i.e., h_i, T_i ($i = 1, 2$) represent the mass thermodynamic state in the control
 198 volume of the FLE.

The water mass and heat balance equations in a FLE control volume V consisting of V_1 and V_2 , can be obtained from mass and enthalpy conservation laws, which read

$$V_1 \dot{w} + A j_h = 0 \quad V_1 \dot{U} + A j_T + V_1 q_T = 0 \quad V_2 \dot{w} - A j_h = 0 \quad V_2 \dot{U} - A j_T + V_2 q_T = 0 \quad (1)$$

199 where the dot accents denote the time rates of variables, $w = w(h, T)$ and $U = U(h, T)$ stand for the

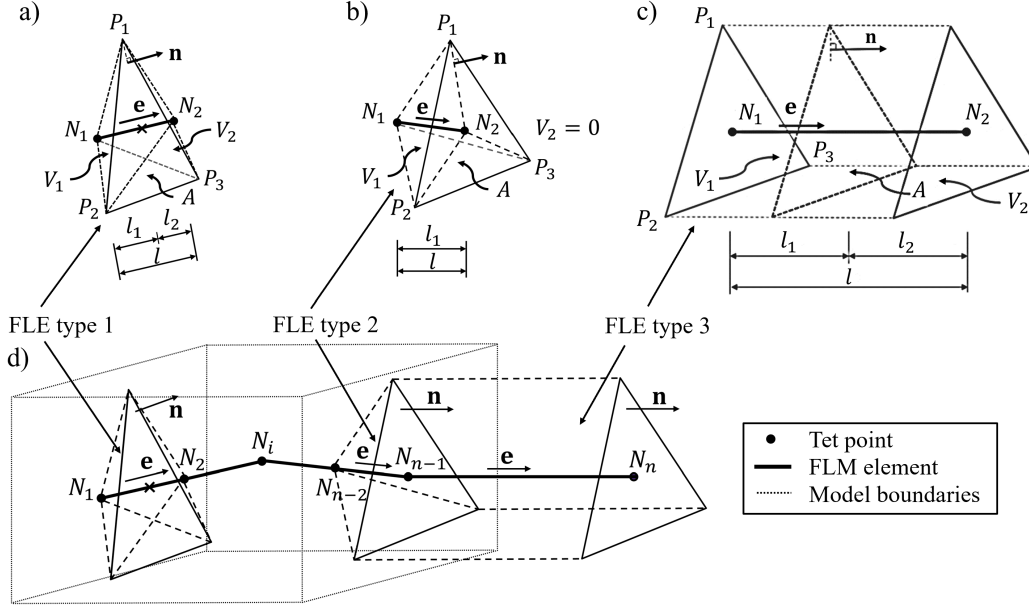


Figure 3: Illustrative diagrams of various *Flow Lattice Element* (FLE) types: a) regular (type 1) element, $V = V_1 + V_2$, $g_1 = l_1/l$, $g_2 = l_2/l$, $\mathbf{e} \nparallel \mathbf{n}$; b) transitional layer (type 2) element, $V = V_1$, $V_2 = 0$, $g_1 = 1$, $g_2 = 0$, $\mathbf{e} \nparallel \mathbf{n}$; c) boundary layer (type 3) element, $V = V_1 + V_2$, $V_1 = V_2$, $g_1 = 1/2$, $g_2 = 1/2$, $\mathbf{e} \parallel \mathbf{n}$; d) FLEs in the model

200 total water mass content and internal energy per unit volume, j_h and j_T stand for the flux density of
 201 water mass and heat per unit time associated with the projected section area A , a positive flux is defined
 202 entering V_1 from V_2 ; q_T stands for the source term of thermal energy. The material volumes V_1 , V_2 , and
 203 their common projected section area A are considered to be constant in this study.

204 In concrete, water is present in multiple phases [67, 68, 69]: (a) capillary water, (b) water vapor, (c)
 205 adsorbed water, and (d) non-evaporable (chemically bound) water. Water that is chemically combined
 206 with other chemical compounds loses its mobility completely and cannot contribute to moisture transport.
 207 In the literature, chemically bound water is typically referred to as non-evaporable water, as it is retained
 208 even at temperatures not exceeding 105 °C. Generally, the transport mechanisms of each phase should be
 209 modeled independently through the formulation of separate diffusion equations, as each mechanism has
 210 its own driving force, e.g. capillary pressure for the capillary water, vapor pressure for the water vapor,
 211 etc.). However, the computational cost of such complex phenomena calls for a simplified analysis. First,
 212 it is possible to simplify the problem by postulating the existence of a local thermodynamic equilibrium
 213 ([69, 70, 18]). This assumption also holds approximately in the presence of chemical evolution of the
 214 system (aging) as the characteristic time scale of the water transport mechanisms is much smaller than
 215 those associated with concrete chemical reactions.

216 The total water content w , can be expressed as $w = w_e + w_n$, where w_e represents the evaporable
 217 water portion, which comprises capillary water, water vapor, and adsorbed water, and w_n represents the
 218 non-evaporable (chemically bound) water [71, 72]. The relation between the evaporable water and relative
 219 humidity h , is the so called *sorption isotherm* which is here assumed also to be function of temperature
 220 T , the degree of hydration, α_c , and the degree of silica-fume reaction, α_s , i.e. $w_e = w_e(h, T, \alpha_c, \alpha_s)$
 221 (age-dependent *adsorption/desorption isotherm* [73]). The water content variation in time reads

$$\dot{w} = \dot{w}_e + \dot{w}_n = \frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \quad (2)$$

222 where $\partial w_e / \partial h$ is the slope of the sorption isotherm (also called *moisture capacity*), $\partial w_e / \partial T$ represents
 223 the variation of the evaporable water with respect to the temperature variation, and the last three
 224 terms express the effect of concrete aging on the diffusion process; the first two terms account for the
 225 microstructural changes associated with the concrete chemical reactions and the last term accounts for
 226 the internal "consumption" of water involved in the chemical reactions, called *moisture sink*, explains the
 227 well-known *self-desiccation* phenomenon.

228 For the temperature range considered in this study (10–100 °C), no phase change is happening in
 229 concrete [74], hence the thermal energy variation in time reads $\dot{U} = \rho c_t \dot{T}$, where ρ is the mass density
 230 and c_t is the isobaric heat capacity (specific heat) of concrete, assumed as $c_t = 1100$ J/kg°C, which may
 231 be generally adopted for all concretes. The product ρc_t is the *thermal capacity*.

232 In this study, all moisture transport mechanisms are combined into a single phenomenological equation
 233 ([69, 75, 76]) and the overall moisture transport process under isothermal conditions is described through
 234 an equivalent Fick's law $j_h = -D_h(h, T)g_h$, where j_h is the flux density of water mass per unit time, g_h is
 235 the discrete estimation of relative humidity gradient, which read $g_h = \mathbf{e} \cdot \mathbf{n}(h_2 - h_1)/l$, the proportionality
 236 coefficient $D_h(h, T)$, called *moisture permeability*, is a nonlinear function of relative humidity h and
 237 temperature T [75]. The choice of relative humidity h as the state variable of the problem (in addition
 238 to the temperature T) is done here for convenience but other options can be equivalently adopted [70].

239 Heat conduction can be described in concrete by classical Fourier's law as $j_T = -\lambda g_T$, where j_T stands
 240 for the heat flux density per unit time, g_T stands for the discrete estimation of the temperature gradient,
 241 which reads $g_T = \mathbf{e} \cdot \mathbf{n}(T_2 - T_1)/l$. For temperatures below 100°C, the heat conductivity of concrete, λ ,
 242 may be assumed constant and a value of $\lambda = 2.3$ W/m°C [74, 77] will be used for the numerical simulations
 243 presented in the following sections.

The source term of thermal energy q_T , can be expressed as $q_T = \dot{Q}_c + \dot{Q}_s$, in which \dot{Q}_c and \dot{Q}_s , represent

the rate of heat generation per unit volume due to cement hydration and silica fume (pozzolanic) reaction, respectively. They read

$$\dot{Q}_c = \dot{\alpha}_c c \tilde{Q}_c^\infty \qquad \dot{Q}_s = \dot{\alpha}_s s \tilde{Q}_s^\infty \qquad (3)$$

244 where $\dot{\alpha}_c$ is the rate of the hydration degree, as discussed following, c is the cement mass content, $\dot{\alpha}_s$
 245 is the rate of pozzolanic reaction degree, also discussed following, and s is the silica fume mass content.
 246 The latent heat of hydration reaction per unit of hydrated mass, \tilde{Q}_c^∞ , may be assumed constant for
 247 a given concrete as it depends only on the chemical composition of its cement [78, 13, 79, 15, 18]. It
 248 can be measured in calorimetric tests at complete hydration and typical values range from 400 kJ/kg to
 249 550 kJ/kg depending upon cement composition [80]. The enthalpy of silica fume, \tilde{Q}_s^∞ , can be assumed
 250 constant and equal to 780 kJ/kg [81].

251 By substituting Eqs. 2, 3 and collecting all the other entries described above, one can expand the
 252 governing equations Eqs. 1 as

$$\begin{aligned} V_1 \left(\frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \right) + AD_h \frac{h_2 - h_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_1 \left(\rho c_t \dot{T} + \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty \right) + A\lambda \frac{T_2 - T_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_2 \left(\frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \right) - AD_h \frac{h_2 - h_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_2 \left(\rho c_t \dot{T} + \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty \right) - A\lambda \frac{T_2 - T_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \end{aligned} \qquad (4)$$

253 3 The Hygro-Thermo-Chemical (HTC) model

254 The HTC model [24, 25], simulating the moisture and temperature evolution in a cementitious material
 255 considering the simultaneous chemical reactions, i.e. hydration of cement and pozzolanic reactions, is
 256 here recalled.

257 3.1 Cement hydration

258 The cement hydration is the result of the reaction between the free water and the cement particles,
 259 which are primarily composed of calcium silicates (C_3S and C_2S) and calcium aluminates (C_3A and
 260 C_4AF). The products of the cement hydration are calcium silicates hydrates (CSH), calcium hydroxide
 261 (CH), ettringite (Aft), and monosulfate (Afm) [82]. Since the precise stoichiometry of the chemical
 262 reactions involved is not exactly known and their mutual interaction, as well as the influence of external

263 factors (such as temperature, relative humidity, etc.), are not completely understood [82, 74], many
 264 researchers [83, 84, 79, 15, 18] have studied cement hydration with reference to an overall hydration
 265 process without distinction among the hydration reactions of every single compound. The present model
 266 also adopts this approach.

267 Following the thermodynamics-based model proposed in [84, 85, 15], the hydration kinetics can be
 268 described by postulating the existence of a Gibb's free energy that is dependent on the external temper-
 269 ature T and the *hydration extent* χ_c . The hydration extent can be used to define a normalized measure
 270 of the hydration process, the *hydration degree*, as $\alpha_c = \chi_c/\bar{\chi}_c^\infty$, in which $\bar{\chi}_c^\infty$ is the theoretical asymptotic
 271 values of the hydration extent in ideal hygrometric conditions. Since the actual asymptotic value of
 272 hydration extent, χ_c^∞ , is always smaller than the theoretical value, $\bar{\chi}_c^\infty$, the ratio between the actual and
 273 the theoretical asymptotic values is the asymptotic degree of hydration $\alpha_c^\infty = \chi_c^\infty/\bar{\chi}_c^\infty$, which has been
 274 shown to be always smaller than unit [80].

275 By assuming that the hydration extent is driven by the thermodynamic force conjugate to it, the so-
 276 called *chemical affinity*, and governed by an Arrhenius-type expression, the following evolution equation
 277 for the hydration degree can be written [84, 15].

$$\dot{\alpha}_c = A_c(\alpha_c) e^{-E_{ac}/RT} \quad \text{with} \quad A_c(\alpha_c) = A_{c1} \left(\frac{A_{c2}}{\alpha_c^\infty} + \alpha_c \right) (\alpha_c^\infty - \alpha_c) e^{-\eta_c \alpha_c / \alpha_c^\infty} \quad (5)$$

278 where $A_c(\alpha_c)$ is the so called *normalized chemical affinity*, E_{ac} is the hydration activation energy, R is
 279 universal gas constant, and η_c , A_{c1} , A_{c2} are material parameters. In the present study the value E_{ac}/R
 280 = 5000 K will be adopted in absence of specific measurements. The parameters η_c , A_{c1} , A_{c2} can be
 281 calibrated by simulating the temperature evolution during adiabatic tests, see [25].

282 Moreover, it is well known from experiments that if the relative humidity decreases below a certain
 283 value ($h \approx 75\%$), the hydration process slows down and may even stop [68, 86]. This phenomenon can
 284 be taken into account [18] by rewriting Eq. 5 as

$$\dot{\alpha}_c = A_c(\alpha_c) \beta_h(h) e^{-E_{ac}/RT} \quad \text{with} \quad \beta_h(h) = [1 + (a - ah)^b]^{-1} \quad (6)$$

285 The function $\beta_h(h)$ is an empirical function with the parameters a and b that may be calibrated through
 286 the analysis of experimental data but constant values of $a = 5.5$ and $b = 4$ are generally adopted [18, 25].

287 3.2 Pozzolanic reaction from silica-fume

288 Silica fume (SF) is often added in the concrete mixture in order to produce HPC [87, 88]. SF belongs
 289 to the category of highly pozzolanic materials and it consists of silica in noncrystalline form with a high
 290 specific surface and exhibiting high pozzolanic activity. The pozzolanic reactions occur between the silica-
 291 rich SF particles and the portlandite (CH) generating Calcium Silicate Hydrates (CSH) [89]. The exact
 292 stoichiometry of pozzolanic reaction between SF and CH is not exactly known, especially in relation to
 293 the water stoichiometry coefficient. As current literature presents dissonant results on this issue, it is
 294 assumed that no water is involved (consumed or released) in the SF pozzolanic reaction [90].

295 The effect of SF on moisture and heat diffusion phenomena is here modeled through the introduction
 296 of the degree of SF reaction, α_s , defined as the ratio between the amount of reacted SF and the total
 297 amount of SF. Since the kinetics of the pozzolanic reaction can be assumed to be a diffusion-controlled
 298 process [91, 92, 93], the theory adopted in the previous section to describe cement hydration can be
 299 exploited here to describe the evolution of SF reaction. Accordingly, one has

$$\alpha_s = A_s(\alpha_s) e^{-E_{as}/RT} \quad \text{with} \quad A_s(\alpha_s) = A_{s1} \left(\frac{A_{s2}}{\alpha_s^\infty} + \alpha_s \right) (\alpha_s^\infty - \alpha_s) e^{-\eta_s \alpha_s / \alpha_s^\infty} \quad (7)$$

300 where A_s is the SF normalized affinity, E_{as} is the activation energy of SF reaction, and α_s^∞ is the
 301 asymptotic value of SF reaction degree. $E_{as}/R = 9700$ K can be generally assumed [80]. The material
 302 parameters A_{s1} , A_{s2} and η_s can be calibrated similarly to the parameters governing cement hydration
 303 [25].

304 When a sufficient amount of SF is available in concrete, all the CH produced by the cement hydration
 305 can be consumed by the pozzolanic reaction. For water-to-cement ration, w/c , of about 0.5 (normal
 306 concrete), about 16% silica fume is required to consume all the calcium hydroxide during the pozzolanic
 307 reaction [94]. For lower values of w/c , the required amount of silica fume to consume all the calcium
 308 hydroxide is reduced proportionally and the proportionality coefficient of 0.4 can be used as confirmed
 309 by experiments on carbonation shrinkage [94]. According to these observations, the asymptotic degree of
 310 SF reaction degree is expressed as

$$\alpha_s^\infty = SF^{eff} \quad \text{if} \quad \begin{cases} s/c < 0.16 & \text{and } w/c > 0.4, \\ s/c < 0.4(w/c) & \text{and } w/c \leq 0.4. \end{cases} \quad (8)$$

311 where s/c is the silica-to-cement ratio and SF^{eff} is the "efficiency" of silica fume, i.e. the mass ratio

312 between the SiO_2 mass content and the total mass of silica fume. Because only the SiO_2 participates in
 313 the pozzolanic reaction, the remaining portion of SF does not participate in any chemical reaction and
 314 has the role of an inert filler. Typically, SF_{eff} ranges from 0.85 to 0.92 [82]. If the amount of silica
 315 fume exceeds the minimum required amount of silica fume to consume all the calcium hydroxide, the
 316 asymptotic degree of SF reaction is less than SF^{eff} . Assuming a linear relationship between α_s^∞ and
 317 s/c , the asymptotic degree of SF reaction degree can be estimated as

$$\alpha_s^\infty = \begin{cases} \frac{SF^{eff} \cdot 0.16}{s/c} & \text{if } w/c > 0.4, \\ \frac{SF^{eff} \cdot 0.4(w/c)}{s/c} & \text{if } w/c \leq 0.4. \end{cases} \quad (9)$$

318 3.3 Asymptotic hydration degree

319 The asymptotic (ultimate) hydration degree, α_c^∞ , is always less than one [80], as a unit value can be
 320 reached only in ideal conditions that in practice are never fulfilled during curing [68].

321 As proposed by Pantazopoulo and Mills [72], α_c^∞ may be calculated on the basis of a mass balance
 322 between water needed to ensure saturation conditions and total water content. This balance gives

$$\alpha_c^\infty = \frac{1.031 w/c}{0.194 + w/c} \quad (10)$$

323 However, the presence of SF further reduces the ultimate achievable hydration degree because, due to
 324 the formation of additional CSH gel from the pozzolanic reaction, additional water is hindered in entering
 325 the gel pores and it becomes unavailable for the cement hydration reaction [80]. As proposed by Di Luzio
 326 and Cusatis [24], α_c^∞ can be expressed in the presence of SF as

$$\alpha_c^\infty = \frac{1.032w/c - 0.279(s/c)\alpha_s^\infty}{0.194 + w/c} \quad (11)$$

327 where s is SF content, and α_s^∞ is the asymptotic SF reaction degree given by Eqs. 8 and 9. Note that
 328 the minus sign in Eq. 11 expresses the fact that in presence of SF the asymptotic degree of hydration
 329 is reduced because the SF pozzolanic reaction increases the specific surface of the CSH gel making
 330 unhydrated cement grain cores less accessible.

331 3.4 Non-evaporable and evaporable water

332 In the present model, the non-evaporable water is the amount of water that is chemically bound as a
 333 consequence of cement hydration and SF reaction, and is computed as

$$w_n(\alpha_c, \alpha_s) = \kappa_c \alpha_c c + \kappa_s \alpha_s s \quad (12)$$

334 where κ_c represents the mass ratio of non-evaporable water at full hydration and κ_s represents the mass
 335 ratio of SF content at full SF reaction.

336 The content of evaporable water is characterized through its relationship with relative humidity, the
 337 so-called "adsorption isotherm" if obtained with increasing relativity humidity, or "desorption isotherm"
 338 in the opposite case. The adsorption and desorption isotherms are typically not the same; a hysteresis is
 339 observed during adsorption/desorption cycles [95, 96]. This has been explained in [97] as a consequence
 340 of two related mechanisms: snap-through instabilities during the filling or emptying of non-uniform
 341 nanopores or nanoscale asperities, and the molecular coalescence, or capillary condensation, within a
 342 partially filled surface. However, in the present model, as a first approximation this difference is neglected,
 343 see also [76], and in the following "sorption isotherm" will be used with reference to both adsorption and
 344 desorption conditions.

345 The shape of the sorption isotherm for concrete is influenced by many factors, particularly those that
 346 influence the extent and rate of the chemical reactions and, in turn, determine pore structure and pore
 347 size distribution. These factors include water to cement ratio, cement chemical composition, silica fume
 348 content, curing time and method, temperature, and mix additives, etc.

349 In the literature, various formulations can be found to describe the sorption isotherm of normal
 350 concrete, see [77, 98]. In the present model, the semi-empirical expression proposed by Norling Mjornell
 351 [73] is adopted because it explicitly accounts for the evolution of hydration reaction and the SF content.
 352 The effect of the temperature on the sorption isotherm is considered by introducing an Arrhenius type
 353 corrective term. This sorption isotherm reads

$$w_e(h, T, \alpha_c, \alpha_s) = G_1(\alpha_c, \alpha_s) \left[1 - \frac{1}{e^{10(g_1 \alpha_c^\infty - \alpha_c)h}} \right] e^{\frac{Q}{R} \left(\frac{1}{T} - \frac{1}{T_0} \right)} + K_1(\alpha_c, \alpha_s) \left[e^{10(g_1 \alpha_c^\infty - \alpha_c)h} - 1 \right] \quad (13)$$

354 where the first term (gel isotherm) represents the physically bound (adsorbed) water, which is corrected
 355 by the Arrhenius term $e^{Q/R(1/T-1/T_0)}$, in order to account for the effect of temperature on the physically
 356 bound water, and the second term (capillary isotherm) represents the capillary water. The temperature

357 effect parameter Q/R is the activation energy, T_0 is the reference room temperature ($T_0 = 296K$). The
 358 material parameter $g_1 > 1$ governs the shape of the sorption isotherm.

359 In Eq. 13 the temperature effect is applied only to the term accounting for the evaporable water in
 360 the gel pores. In compliance with the conclusions drawn by several authors [99, 100, 101], the model
 361 aims to describe the phenomenon for which, under the same conditions, increasing temperature hinders
 362 adsorption and promotes desorption, resulting in a fewer number of water molecules that can be adsorbed.
 363 The opposite process is instead observed when the temperature decreases, i.e. more evaporable water is
 364 available under the same condition at a lower temperature.

365 The coefficient $G_1(\alpha_c, \alpha_s)$ represents the amount of water per unit volume held in the gel pores at
 366 100% relative humidity, and it can be expressed [73] as

$$G_1(\alpha_c, \alpha_s) = k_{vg}^c \alpha_c c + k_{vg}^s \alpha_s s \quad (14)$$

367 where k_{vg}^c and k_{vg}^s are material parameters.

368 The maximum amount of water per unit volume, $w_e(h = 1, T, \alpha_c, \alpha_s)$, that can fill pores (both
 369 capillary pores and gel pores) at a certain degree of hydration under saturation conditions is given
 370 $w_0 - 0.188c\alpha_c + 0.22s\alpha_s$. By using this expression and the Eq. 13 with $h = 1$, and solving for $K_1(T, \alpha_c, \alpha_s)$
 371 one obtains

$$K_1(T, \alpha_c, \alpha_s) = \frac{w_0 - 0.188c\alpha_c + 0.22s\alpha_s - G_1 [1 - e^{-10(g_1\alpha_c^\infty - \alpha_c)}] e^{\frac{Q}{R}(\frac{1}{T} - \frac{1}{T_0})}}{e^{10(g_1\alpha_c^\infty - \alpha_c)} - 1} \quad (15)$$

372 The material parameters k_{vg}^c , k_{vg}^s , and g_1 can be calibrated by fitting experimental data relevant to free
 373 (evaporable) water content in concrete at various ages [73, 25].

374 **3.5 Moisture permeability**

375 Moisture transport in concrete depends on various diffusion mechanisms that characterize the mobility
 376 of different water phases present in concrete, and which are influenced by the pore structure of cement
 377 paste, such as molecular diffusion (ordinary diffusion), Knudsen diffusion, and surface diffusion [102,
 378 103]. Although each individual mechanism is reasonably understood, it is not always easy to make an
 379 accurate prediction of the total diffusivity; it is difficult to describe and simulate the details of concrete
 380 pore structure formed by randomly oriented pores with variable pore radii and with varying degrees of
 381 tortuosity and connectivity. In addition, the scenario is further complicated by the dependence of the

382 pore structure on water to cement ratio, cement content, silica fume content, age, etc.

383 In this work, as typically done in literature [75], the above mechanisms are lumped and described phe-
384 nomenologically through equivalent Fick’s law (see Sec 2.2), characterized by the moisture permeability
385 given by

$$D_h(h, T) = \psi(T)D_1 \left[1 + \left(\frac{D_1}{D_0} - 1 \right) (1 - h)^n \right]^{-1} \quad \text{with} \quad \psi(T) = \exp \left(\frac{E_{ad}}{RT_0} - \frac{E_{ad}}{RT} \right) \quad (16)$$

386 In Eq. 16, the coefficient $\psi(T)$ takes into account the effect of temperature on permeability [104]
387 as earlier proposed by Bažant [75], D_0 and D_1 represent moisture permeability for a completely dry
388 situation ($h = 0$) and completely saturated situation ($h = 1$), respectively. The exponent n governs the
389 rate at which permeability transitions from D_0 to D_1 . In Eq. 16, T is the absolute temperature, T_0 is
390 the reference room temperature ($T_0 = 296K$) and $E_{ad}/R \approx 2700K$ [75]. The material parameters D_0 ,
391 D_1 , and n can be calibrated by fitting experimental data relevant to moisture diffusion [25].

392

393 3.6 HTC formulation at the mesoscale

394 The heterogeneity considered in the FLM is the meso-scale heterogeneity related to the resolution
395 scale of LDPM. Although, some authors use the term “meso-scale” in a wider sense to include concrete
396 models targeting concrete a smaller scale. In FLM the 1D conduits are located and oriented according to
397 the cement paste matrix distribution. This feature reflects ipso facto the discrepancy in the permeability
398 of impermeable aggregates and permeable cement paste matrix. The FLM can capture the tortuosity of
399 the flow network in the cementitious material given by the topological distribution of coarse aggregates.
400 This means that FLM is able to account for the material heterogeneity but only at the mesoscale level,
401 same scale (resolution) of LDPM. If the heterogeneities associated of smaller scale properties would be
402 taken into account, they must be included in the HTC constitutive laws.

403 The implementation of the HTC model in the 3D FLM system paves the way for a new version of the
404 formulation that can better exploit the features of the discrete mesoscale implementation. An update
405 HTC formulation could be done by scaling all the properties (material parameters) based on the real
406 cement paste content in the volume in each conduit (Flow Lattice Element). Another update would be
407 a new calibration of the moisture diffusion parameters (D_0 , D_1 , and n) in order to reflect the mesoscale
408 geometry (heterogeneity) of the fluid flow network after considering all the coarse aggregate particles in

409 the construction of the LDPM geometry. However, those updates and improvements are behind the scope
 410 of the manuscript and are left for future work.

411 The HTC formulation could be also improved by enhancing the mesoscale heterogeneity, which is
 412 considered in FLM since it is related to the resolution scale of LDPM, i.e. length scale 10^{-3} m. The
 413 heterogeneity can be extend at smaller scale, e.g. [105, 106], by a multi-scale approach by linking to
 414 the material porosity the HCT constitutive laws. However, this feature is also behind the scope of the
 415 manuscript.

416 4 Numerical implementation

417 In this work, the authors utilized a commercial finite element software *Abaqus* as the solver for the
 418 nonlinear system of equations, through the Abaqus user element subroutines for implicit analyses, named
 419 *UEL*. Generally, for a *UEL*, users should calculate the linearized governing equations (incremental form)
 420 for the nonlinear problems, by providing the Jacobian (a.k.a. the tangent stiffness) matrix *AMATRIX*
 421 and the right hand side (a.k.a. the residual) vector *RHS* of the incremental governing equations to the
 422 Abaqus solver, to calculate the nodal increments of degrees of freedom for user elements during each
 423 iteration, until the convergence is reached. The remaining parts, such as matrix assemblage and matrix
 424 solving, are automatically processed by the Abaqus solver for implicit analyses *Abaqus/Standard*.

425 To calculate the incremental form for the HTC model within the FLE, one may rewrite the governing
 426 equations for the FLE, Eqs. 4, in a matrix form as

$$\mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} = \mathbf{0} \quad (17)$$

427 where,

$$\mathbf{M} = V \begin{bmatrix} g_1 C_1 & 0 & 0 & 0 \\ 0 & g_1 C_3 & 0 & 0 \\ 0 & 0 & g_2 C_1 & 0 \\ 0 & 0 & 0 & g_2 C_3 \end{bmatrix} \quad (18)$$

$$\mathbf{K} = \frac{A}{l} \begin{bmatrix} D_h & 0 & -D_h & 0 \\ 0 & \lambda & 0 & -\lambda \\ -D_h & 0 & D_h & 0 \\ 0 & -\lambda & 0 & \lambda \end{bmatrix} \quad (19)$$

$$\mathbf{S} = V \begin{bmatrix} g_1 C_2 & g_1 C_4 & g_2 C_2 & g_2 C_4 \end{bmatrix}^T \quad (20)$$

$$\mathbf{u} = \begin{bmatrix} h_1 & T_1 & h_2 & T_2 \end{bmatrix}^T \quad (21)$$

431 in which, g_1 and g_2 are the length proportionality coefficients (see Sec. 2.1), the volumes are $V_1 = Vg_1$ and
 432 $V_2 = Vg_2$; the coefficients, which are estimated by the weighted averages of internal variables, read $D_h =$
 433 $D_h (h_1 g_2 + h_2 g_1, T_1 g_2 + T_2 g_1)$, $C_1 = \partial w_e / \partial h = g_2 (\partial w_e / \partial h)|_1 + g_1 (\partial w_e / \partial h)|_2$, $C_2 = g_2 (\dot{q}_h)|_1 + g_1 (\dot{q}_h)|_2$,
 434 $C_3 = \partial U / \partial T = \rho c_t = g_2 (\rho c_t)|_1 + g_1 (\rho c_t)|_2$, and $C_4 = g_2 (\dot{q}_T)|_1 + g_1 (\dot{q}_T)|_2$ where $\dot{q}_h = \partial w_e / \partial \alpha_c \cdot \dot{\alpha}_c +$
 435 $\partial w_e / \partial \alpha_s \cdot \dot{\alpha}_s + \dot{w}_n$ and $\dot{q}_T = \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty$. The subscripts 1 and 2 stand for the variables evaluated
 436 at node N_1 and N_2 , respectively.

437 The above matrix form of the governing equations is nonlinear and is hard to be solved numerically.
 438 To linearize the nonlinear equations, one may rewrite Eq. 17 as: $\mathbf{f}(\mathbf{u}) = \mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} = \mathbf{0}$. By adopting
 439 a root-finding algorithm, such as Newton-Raphson method (a.k.a. Newton's method) in this work, one
 440 may linearize the problem and approximate the nonlinear governing equations with the incremental form,
 441 which is given by

$$\mathbf{f}(\mathbf{u}_{n+1}) \approx \mathbf{f}(\mathbf{u}_n) + \frac{\partial \mathbf{f}(\mathbf{u}_n)}{\partial \mathbf{u}} \Delta \mathbf{u} = \mathbf{0} \quad (22)$$

$$\frac{\partial \mathbf{f}(\mathbf{u}_n)}{\partial \mathbf{u}} \Delta \mathbf{u} = -\mathbf{f}(\mathbf{u}_n)$$

442 where the subscripts n and $n + 1$ stand for the current and next iteration step for Newton's method

443 respectively, and

$$\mathbf{f}(\mathbf{u}) = \begin{bmatrix} f_1 \\ f_2 \\ f_3 \\ f_4 \end{bmatrix} = \begin{bmatrix} Vg_1C_1\dot{h}_1 + \frac{A}{l}D_h(h_1 - h_2) + Vg_1C_2 \\ Vg_1C_3\dot{T}_1 + \frac{A}{l}\lambda(T_1 - T_2) + Vg_1C_4 \\ Vg_2C_1\dot{h}_1 - \frac{A}{l}D_h(h_1 - h_2) + Vg_2C_2 \\ Vg_2C_3\dot{T}_1 - \frac{A}{l}\lambda(T_1 - T_2) + Vg_2C_4 \end{bmatrix} \quad (23)$$

444 The multiplier on the left-hand side of Eq. 22, $\partial \mathbf{f}(\mathbf{u}_n) / \partial \mathbf{u}$, is corresponding to the Jacobian matrix
 445 *AMATRIX* and the term on the right-hand side $-\mathbf{f}(\mathbf{u})$ is the right-hand side vector *RHS* in the Abaqus
 446 user element. The entries of the Jacobian matrix and the right-hand side vector corresponding to Eqs.
 447 22 and 23 can be calculated by

$$\frac{\partial \mathbf{f}(\mathbf{u}_n)}{\partial \mathbf{u}} = \begin{bmatrix} \frac{\partial f_1}{\partial h_1} & 0 & \frac{\partial f_1}{\partial h_2} & 0 \\ 0 & \frac{\partial f_2}{\partial T_1} & 0 & \frac{\partial f_2}{\partial T_2} \\ \frac{\partial f_3}{\partial h_1} & 0 & \frac{\partial f_3}{\partial h_2} & 0 \\ 0 & \frac{\partial f_4}{\partial T_1} & 0 & \frac{\partial f_4}{\partial T_2} \end{bmatrix} \quad (24)$$

448 and

$$-\mathbf{f}(\mathbf{u}) = - \begin{bmatrix} f_1 & f_2 & f_3 & f_4 \end{bmatrix}^T \quad (25)$$

449 The linearized governing equations are then assembled for all FLE control volumes and the assembly
 450 must be completed by appropriate boundary and initial conditions.

451 4.1 Time integration scheme

452 The backward Euler method is used in Abaqus/Standard for the time integration; it is unconditionally
 453 stable and allows large time increments. For the transient analysis of the HTC model, the backward Euler
 454 method is considered very effective, as the total simulation time is always extensive. Specifically, one can
 455 rewrite the governing equations 17 as

$$\mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} \approx \mathbf{M}_{n+1} \frac{\mathbf{u}_{n+1} - \mathbf{u}_n}{\Delta t_{n+1}} + \mathbf{K}_{n+1}\mathbf{u}_{n+1} + \mathbf{S}_{n+1} = \mathbf{0} \quad (26)$$

456 In which the subscripts n and $n + 1$ here represent the previous time step and the current time step. One
 457 may notice that the incremental form of the governing equations should take the partial derivatives at

458 the current time and that the matrices and the known term must be evaluated at the current time. The
459 root-finding technique, i.e. the Newton-Raphson method, is then used for solving Eq. 22 for the FLM
460 implementation.

461 Additionally, numerical implementation of the formulation in the previous sections requires, at the
462 time integration of the internal variables, hydration degree and silica-fume reaction degree at each tet
463 point of the flow lattice element mesh. The governing equations of the internal variables (Eqs. 5 and
464 7) can be all expressed in the form $\dot{x} = f(x, h, T)$, where x represent the internal variable. A very
465 effective method to integrate these internal variables is the use of the second order Runge-Kutta formula
466 (midpoint method) by which the value x_{n+1} of the internal variable at time t_{n+1} is given by $x_{n+1} \approx$
467 $x_n + \Delta t_n f(x_{n+1/2}, h_{n+1/2}, T_{n+1/2})$; where x_n is the value of the internal variable at time t_n , Δt_n is the
468 time increment, $x_{n+1/2} = (x_{n+1} + x_n)/2$, $h_{n+1/2} = (h_{n+1} + h_n)/2$, and $T_{n+1/2} = (T_{n+1} + T_n)/2$. All
469 internal variables have zero value as an initial condition at the time $t_0 = 0$ (time of casting).

470 4.2 Boundary conditions

471 On the model boundaries, the mass and heat exchange between the concrete surface and the en-
472 vironment may be considered dependent not only on the conduction/diffusion, but also on the other
473 mechanisms such as the heat convection and the phase change of water mass. In this study, one as-
474 sumes the mass and heat exchange to be linearly dependent on the difference of surface relative humidity
475 and environmental relative humidity, and the difference between surface temperature and environmental
476 temperature, respectively.

477 A "boundary layer" (see Fig. 3) consisting of boundary layer (type 3) FLEs, as mentioned previously,
478 extruding outwards from the concrete surface with prismatic volumes $V_1 = V_2 = V/2$, constant cross-
479 section A , and lengths $l_1 = l_2 = l/2$, is added to the FLE network to simulate the water mass and
480 heat exchange between the specimen and the environment at the specimen surface. All boundary layer
481 FLEs are assumed to have the same material parameters of the FLEs which share the nodes on the
482 specimen surface. In this way, the boundary conditions applied on the boundary layer are simplified to
483 Dirichlet type conditions, $h_{bo} = h_{en}$, and $T_{bo} = T_{en}$, where, h_{bo} and T_{bo} are the relative humidity and
484 the temperature of the extended boundary layer nodes; h_{en} and T_{en} are the relative humidity and the
485 temperature of the environment. By varying the length of the boundary layer, one can simulate various
486 degrees of surface emissivity.

5 Numerical simulations and comparisons with experimental data

5.1 Concrete drying

The first benchmark case for the calibration and validation of FLM focuses on the evolution of hydration degree and the spatial gradients of temperature and humidity fields in the case of the concrete moisture drying at early stages. The capability of FLM to capture the spatial gradients of temperature and humidity fields, as well as the capability of simulating the evolution of the hydration degree, is investigated by comparing FLM results with the reference experimental data of Kim and Lee [10]. The results were also compared with the HTC finite element implementation presented in Di Luzio and Cusatis [25].

Drying tests were conducted to observe the outward water mass flux at the boundaries due to the exposure of the concrete specimen to the ambient air with a smaller relative humidity, and the moisture diffusion inside the specimen due to the imbalance of relative humidity distribution from the core to the external surfaces. Self-desiccation also influences the entirety of the concrete drying process, predominantly at early ages.

The drying test settings in Kim and Lee [10] are as illustrated in Fig. 4: prismatic concrete samples prepared with three mixes were exposed to ambient air with 50% relative humidity at one side, all other sides sealed to create an uniaxial moisture diffusion condition. The three mixes are categorized by their water-to-cement ratios, 0.28, 0.4, and 0.68. The corresponding mix compounds and concrete compositions for each category are found in [10]. The concrete specimens were moist-cured before drying, and the moisture distributions of specimens were measured at the sections spacing 3, 7, and 12 cm from the exposure surface, at two ages $t_0 = 3$ days and $t_0 = 28$ days. In addition to the drying specimens, variation in relative humidity due to self-desiccation only was also investigated via fully sealed concrete cubes with the same three mixes.

The continuous FEM simulations were performed with the finite element mesh as shown in Fig. 4a and FLM simulations with the mesh shown in Fig. 4b. It should be noticed that a boundary layer of FEM elements and FLE elements with the layer thickness of 1 mm was applied to the exposed surfaces of the respective models to properly reproduce the experimental conditions. The model parameters were adopted from [25], as these parameters captured accurately the self-desiccation experiments of water-cement-ratio $w/c = 0.4$ and age $t_0 = 3$ and $t_0 = 28$ days, as well as the drying experiments of water-cement-ratio $w/c = 0.4$ and age $t_0 = 3$ days. Table 1 summarizes the full list of model parameters used in the FLM

517 simulations.

518 Fig. 5a and b report the evolution of relative humidity vs. age for sealed concrete subjected to self-
519 desiccation only after an initial curing of $t_0 = 3$ days and $t_0 = 28$ days. The results show that the
520 relative humidity dropped after the initial curing, indicating that the water provided by moist-curing
521 was not sufficient to maintain a fully saturated state. Furthermore, the concrete specimens with a low
522 water-to-cement ratio tended to have a more significant drop in relative humidity, as expected.

523 The agreement of FLM simulation results with those of continuous FEM simulations and experimental
524 data show that FLM can simulate accurately the drying test experiments. As required inputs for the
525 drying test simulations, the initial hydration degree for mixes $w/c = 0.28$, $w/c = 0.4$, $w/c = 0.68$ due
526 to self-desiccation were recorded as 0.3166, 0.3666, and 0.4323 at $t_0 = 3$ days and 0.4721, 0.5444, and
527 0.6376 at $t_0 = 28$ days, respectively. The initial hydration degree α_{c0} values are also reported in Tab. 1.

528 Fig. 6a, b and c show the experimental and numerical results of the relative humidity at multiple
529 distances from the exposed surface for the drying tests, for an initial curing $t_0 = 3$ days. One can
530 observe from the moisture transport in the concrete specimens at different ages that FLM captured
531 well the gradient of relative humidity for various mixes by comparing to the experimental data and FEM
532 simulation results. Moreover, the concretes with a low water-to-cement ratio dried quicker when compared
533 to ones with higher water-to-cement ratios. The high initial relative humidity for high water-to-cement
534 ratio concretes and the slightly lower initial relative humidity for low water-to-cement ratio concrete also
535 suggest the correct modeling of initial curing for these drying test specimens. Similar observations can
536 be concluded from the experimental and numerical results for drying tests of different mixes with initial
537 curing $t_0 = 28$ days in Fig. 7a, b and c.

538 By comparing the results of self-desiccation tests and drying tests in Fig. 5, Fig. 6 and Fig. 7, one
539 observes that self-desiccation is the predominant cause of the decrease in relative humidity when the
540 water-to-cement ratios of specimens are low, and moisture transport is the predominant cause when
541 water-to-cement ratios are high. This observation is in accordance with known behavior, where concrete
542 with low water-to-cement ratios tends to form a denser microstructure and consequently consumes more
543 water during hydration. The overall results of FLM simulations validate the capability of FLM for
544 capturing self-desiccation, as well as the moisture transport procedure in various concrete mixes.

Table 1: Model parameters used in numerical simulations in Section 5

Parameter [unit]	Section 5.1 and 5.2	Section 5.3
ρ [kg/m ³]	2500	2400
w/c [-]	0.28/0.40/0.68	0.25
c [kg/m ³]	541/423/310	748
κ [W/m/K]	2.5	2.5
c_t [J/K/kg]	1100	1100
A_{c1} [1/s]	4166.5	55600.0
A_{c2} [-]	5.00E-02	1.00E-04
η_c [-]	8.0	8.0
E_{ac}/R [K]	5000	5000
a [-]	5.5	5.5
b [-]	4.0	4.0
\tilde{Q}_c^∞ [J/kg]	5.00E+05	5.00E+05
D_0 [kg/m/s]	2.80E-10/6.39E-10/2.30E-09	6.00E-10
D_1 [kg/m/s]	3.73E-08/7.12E-08/1.97E-07	1.00E-07
n [-]	3.0	4.0
E_{ad}/R [K]	2700	840
T_{ref} [°C]	23.00	20.00
k_{vg}^c [-]	0.255	0.220
g_1 [-]	1.2	1.1
κ_c [-]	0.253	0.185
s [kg/m ³]	0.0	224.3
\tilde{Q}_s^∞ [J/kg]	0.0	7.80E+05
E_{as}/R [K]	0.0	9700
A_{s1} [1/s]	0.0	1.39E+10
A_{s2} [-]	0.0	1.00E-06
η_s [-]	0.0	9.5
k_{vg}^s [-]	0.0	0.36
SF^{eff} [-]	0.0	1.0
α_{c0} [-] (initial hydration degree)	3days 0.3166/0.3666/0.4323 28days 0.4721/0.5444/0.6376	0.0
Q/R [K]	0.0	1500

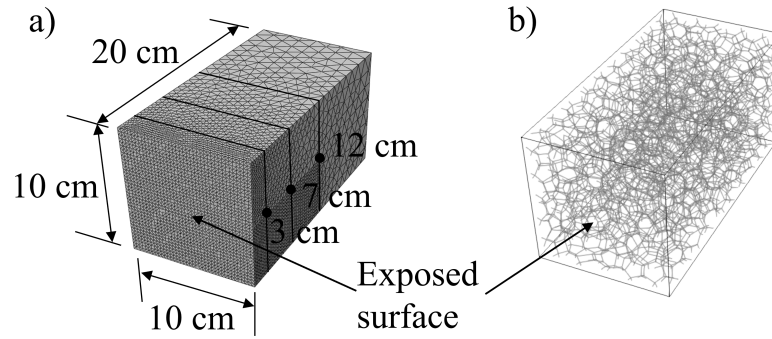


Figure 4: Experimental and numerical setups of drying tests of Kim and Lee [10]: a) dimensions of the specimen and the finite element mesh, b) FLM mesh

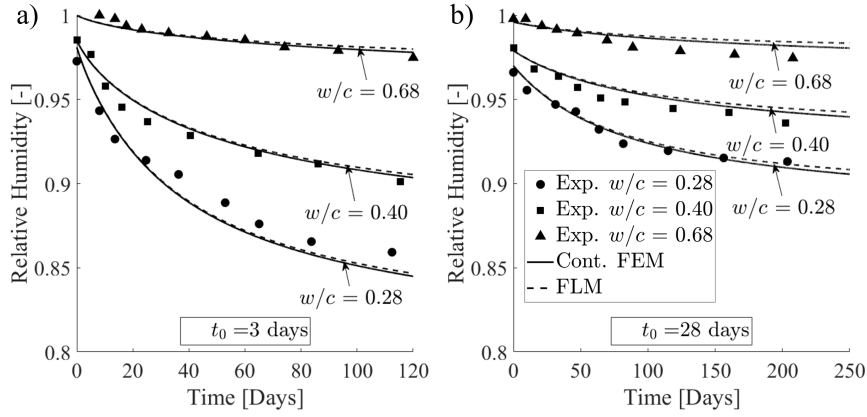


Figure 5: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age due to self-desiccation only, for a) $t_0 = 3$ days, and b) $t_0 = 28$ days

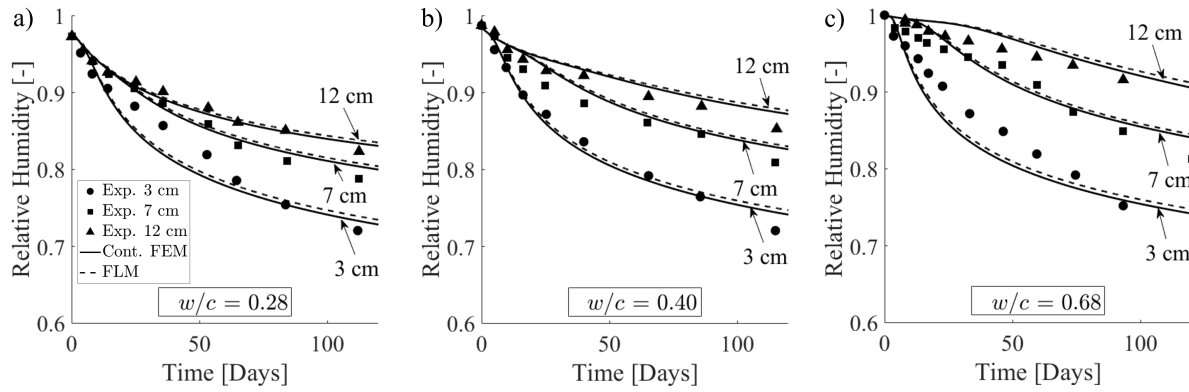


Figure 6: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age at different sections from the exposed surface of concrete specimens due to drying and self-desiccation for $t_0 = 3$ days, with a) mix $w/c = 0.28$, b) mix $w/c = 0.40$, and c) mix $w/c = 0.68$

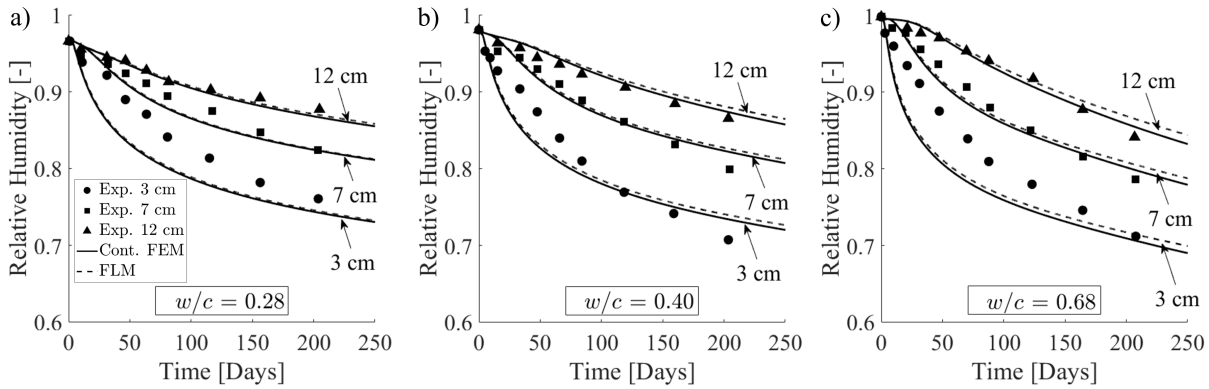


Figure 7: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age at different sections from the exposed surface of concrete specimens due to drying and self-desiccation for $t_0 = 28$ days, with a) mix $w/c = 0.28$, b) mix $w/c = 0.40$, and c) mix $w/c = 0.68$

5.2 Temperature-dependent desorption isotherm

In the previous section, the overall agreement regarding the water diffusion, as well as self-desiccation results with experimental data, show that FLM has the capability to capture moisture transport in drying tests of concrete with various water-to-cement ratios. Yet, the verification of FLM to capture hydration evolution and moisture transport under different temperatures is still needed. In this section, the formulation of the temperature-dependent sorption isotherm presented in the previous Sec. 3.4 is first verified. For this purpose, the desorption tests of 3.6 years-old mortar ($w/c = 0.50$) with reference to the experimental data of Hundt and Kantelberg [107] are here considered, to calibrate the activation energy Q/R within the temperature dependent term in Eq. 13.

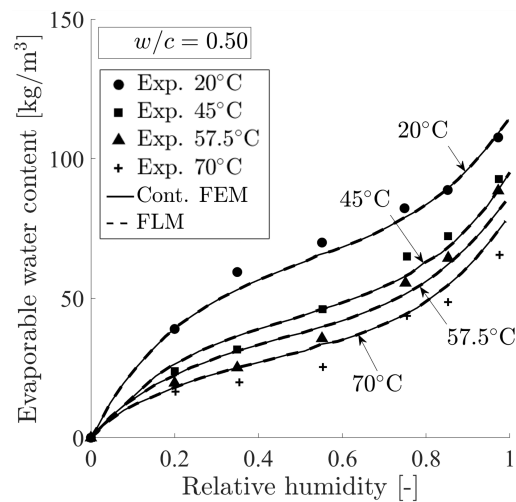


Figure 8: Calibration of the sorption law against the experimental isotherms of Hundt and Kantelberg [107]: evaporable water content vs. relative humidity at various room temperatures, in mortar specimens with mix $w/c = 0.50$

The calibration is accomplished by looking for the best match between the experimental isotherms and the numerical simulation results returned by the employment of the HTC sorption law in Eq. 13. The good agreement between numerical and laboratory outcomes is achieved with $Q/R = 1500K$, see Fig. 8. The model capability of capturing the experimental trends demonstrates the effectiveness of the approach adopted to account for the temperature effect on the variation of the evaporable water content.

5.3 Concrete drying at early-age

In this section drying tests on Ultra High Performance Concrete (UHPC) with reference to the experimental data of Soliman [108] are considered.

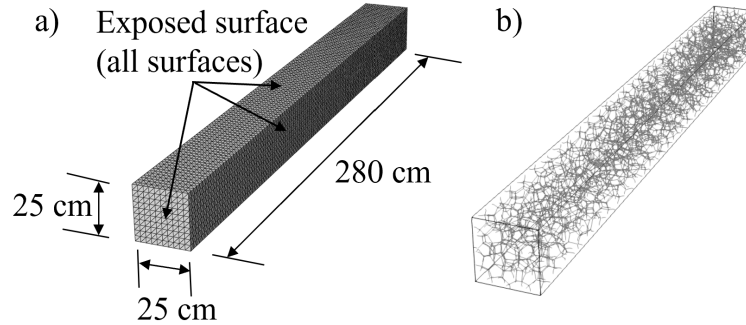


Figure 9: Experimental and numerical setups of drying tests of Soliman [108]: a) dimensions of the specimen and the finite element mesh, b) FLM mesh

562 The specimen settings in Soliman [108] are, as illustrated in Fig. 9a. Prismatic concrete specimens
 563 with dimensions 25 mm \times 25 mm \times 280 mm were dried at all surfaces exposed to the ambient conditions.
 564 All specimens were taken from a single batch and, after casting, were maintained at a room temperature
 565 ($20 \pm 1^\circ\text{C}$) and covered with polyethylene sheets until demolding to avoid moisture loss. All specimens
 566 were demolded at the final setting time and initial readings were taken before moving specimens to the
 567 pre-determined curing conditions: 10°C , 20°C , and 40°C , and ambient air with 40%, 60%, and 80%
 568 relative humidity. All surfaces of the specimens were at the same temperature and relative humidity. In
 569 the experimental investigation the authors measured the mass loss of the specimens and the evolution
 570 of hydration degree under sealed conditions at different temperatures. The investigated mix features a
 571 water-to-cement ratio w/c equal to 0.25; the corresponding mix design and concrete composition is found
 572 in [108] and also listed in Tab. 1.

573 The FEM mesh and FLM mesh are shown in Fig. 9a and b, respectively. Exposed surfaces (all
 574 surfaces) of the respective models were covered by a boundary layer of FEM and FLM elements with
 575 the thickness of 1 mm. Model parameters for hydration were firstly calibrated using the experimental
 576 data from sealed tests at 10°C , 20°C , 40°C . Then, the permeability parameters were calibrated from the
 577 drying tests at temperatures 20°C and 40°C with ambient air relative humidity of 60%. The full list of
 578 model parameters used in the simulations in this section can be found in Tab. 1. All remaining curves
 579 are model predictions.

580 Fig. 10 reports the evolution of the hydration degree vs. age for experiments and numerical sim-
 581 ulations. The agreement between numerical results and experimental data shows that FLM simulated
 582 the cement hydration behaviors of UHPC accurately. The numerical results of the water loss tests are
 583 reported in Fig. 11 showing an overall sufficient fitting of the experimental data. Some differences appear

584 in the simulations of the tests at 10° C, and at 40° C with 80% relative humidity, although by means of
 585 the Arrhenius type corrective terms, the HTC formulation appears to have the capability of capturing
 586 experimental trends.

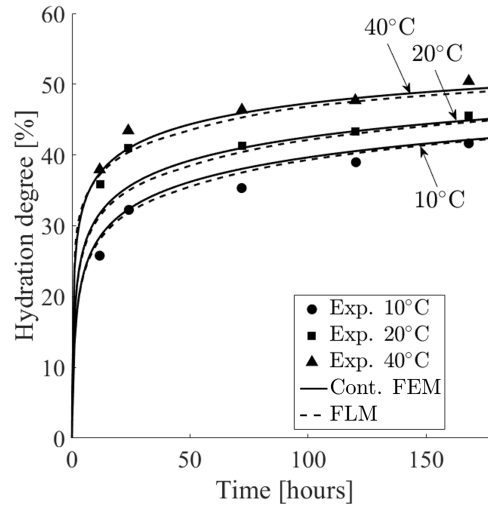


Figure 10: Numerical simulations of the experimental data of Soliman [108]: hydration degree vs. age due to self-desiccation only, in concrete specimens with mix $w/c = 0.25$ at various room temperatures

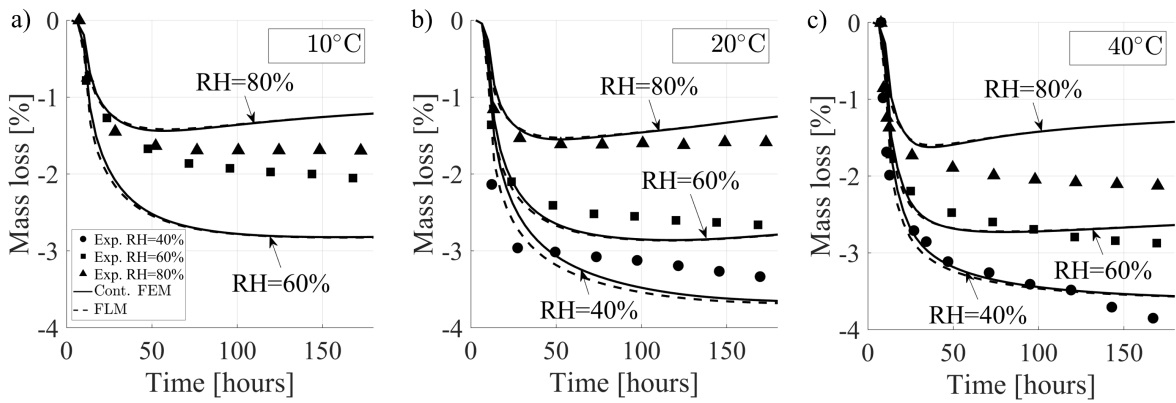


Figure 11: Numerical simulations of the experimental data of Soliman [108]: water loss vs. age due to drying and self-desiccation in concrete mix $w/c = 0.25$ specimens at various ambient temperatures, a) $T = 10^\circ$, b) $T = 20^\circ$, and c) $T = 40^\circ$

587 By increasing the temperature from 10° C to 40° C, competing phenomena are triggered. On one
 588 hand, the hydration process is accelerated, thus self-desiccation is increased as less water available for
 589 the diffusion. On the other hand, at higher temperatures fewer water molecules are adsorbed by the
 590 walls of the gel pores, resulting in lower availability of water for feeding the aging reactions. In this
 591 case, a lower consumption of water is expected. The process is further complicated by drying due to

592 the imbalance between the inner moisture content and the environmental relative humidity, a process
593 which is highly dependent on temperature. The experiments show that moisture loss increases when
594 temperature increases, particularly in drier environments (i.e. 60% and 40% in Fig. 11). This trend
595 is less clear when material aging occurs at higher relative humidity (i.e. 80% in Fig. 11). The model
596 simulates with sufficient accuracy the material behavior at 20° C and 40° C, whereas it overestimates the
597 variation in moisture loss between the responses at 10° C with relative humidity equals to 80% and 60%.
598 This discrepancy between the experimental study and the numerical methods can be explained by the
599 fact that the temperature dependence of the sorption isotherm and of the diffusion coefficient (for which
600 no experimental data were available for this experimental data [108]) seems not capturing very well the
601 temperature variations being the tests at 20° C simulated sufficiently well. Without a doubt, this feature
602 of model needs further future validation with a comprehensive experimental data in which additional
603 data would be available (e.g. sorption isotherm and diffusion coefficient at different temperatures) not
604 only the hydration evolution at different temperatures as for [108].

605 6 Conclusions and future work

606 In this study, a 3D Flow Lattice Model (FLM) with the application to the Hygro-Thermo-Chemical
607 (HTC) model [24, 25] has been formulated, for the simulation of moisture transport and heat transfer in
608 cementitious materials. FLM utilized the topology of the Lattice Discrete Particle Model (LDPM) for
609 the construction of the Flow Lattice Element (FLE) network, and the governing equations of FLE have
610 been derived through the mass and enthalpy conservation laws. The numerical implementation of FLM
611 is implemented via Abaqus user element subroutines *UEL*.

612 The implemented FLM has been validated using a large set of experimental data available in the
613 literature. The results show the following conclusions:

- 614 1. FLM can simulate and predict with high accuracy the moisture variation in self-desiccation and
615 drying experiments for a wide range of standard and high-performance concrete mixes.
- 616 2. The FLM implementation of HTC sorption laws, enriched with the Arrhenius temperature-dependent
617 term, can simulate with sufficient accuracy the effect of temperature on the variation of the evap-
618 orable water content.
- 619 3. Implementation in the LDPM framework in part develops an enhanced Multiphysics-Lattice Dis-
620 crete Particle Model (M-LDPM) framework which may be used for computational analysis of frac-

621 ture permeability behavior of cementitious materials.

622 FLM, combined with the dual lattice topology, enables the seamless coupling of the mechanical and
623 diffusion/chemical behavior of the material at the aggregate scale. As a consequence, the variation of
624 the conductivity/permeability induced by fracturing processes can be simulated by relating the transport
625 properties of flow elements with local cracking behaviors. However, this is beyond the purpose of this
626 paper and it will be discussed in subsequent work.

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631 access-plan](http://energy.gov/downloads/doe-public-access-plan)).

632 The authors confirm that the data supporting the findings of this study are available within the article
633 and its supplementary materials.

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